

$$z_{\text{tot}} = \frac{2}{3} \frac{1}{m_{\text{Ag}}} M_{\text{B}} \left(\frac{dB}{dz} \right) \cdot t = 1.79 \times 10^{-25} \text{ kg}$$

$$\left(\frac{dB}{dz} \right) = \frac{2}{3} \frac{m_{\text{Ag}}}{z_{\text{tot}}} \cdot \frac{1}{M_{\text{B}}} \cdot \frac{1}{t^2}$$

$$= \frac{2}{3} \cdot 1.79 \cdot 10^{-25} \text{ kg} \cdot 5 \cdot 10^{-4} \text{ m} \cdot \frac{1}{9.27 \cdot 10^{-4} \frac{\text{J}}{\text{T}}} \cdot (250 \text{ s}^{-1})^2$$

$$= 0.805 \text{ T/m}$$

7-44) a) CARBON $Z=6$ $(1s)^2(2s)^2(2p)^2$

b) OXYGEN $Z=8$ $(1s)^2(2s)^2(2p)^4$

c) ARGON $Z=18$ $(1s)^2(2s)^2(2p)^6(3s)^2(3p)^6$

$$7-48) E_n = - \frac{Z_{\text{eff}}^2 \cdot E_1}{n^2}$$

$$Z_{\text{eff}} = n \sqrt{\frac{-E_n}{E_1}} = 3 \sqrt{\frac{5.14 \text{ eV}}{13.6 \text{ eV}}} = 1.84$$

7-55) $\Delta l = \pm 1$ $\Delta j = \pm 1, 0$

	Δl	Δj	
$4s_{1/2} \rightarrow 3s_{1/2}$	0	0	FORBIDDEN

$4s_{1/2} \rightarrow 3p_{3/2}$	1	1	ALLOWED
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$4p_{3/2} \rightarrow 3s_{1/2}$	-1	-1	ALLOWED
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